

SCIENTIFIC INSIGHTS

Aliphatic Rings as Bioisosteres of Phenyl Ring in Drug Discovery

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Introduction

The phenyl ring is one of the most prevalent structural motifs in marketed drugs. However, its inherent aromaticity and potential for metabolic oxidation, poor solubility and low permeability can pose challenges in terms of stability and pharmacokinetics. Aliphatic rings, on the other hand, offer improved metabolic stability, reduced lipophilicity, increased solubility and enhanced membrane permeability, making them attractive alternatives to the phenyl ring (**Figure 1**). ^[1] By replacing the phenyl moiety with aliphatic rings, medicinal chemists can modulate the physicochemical properties of compounds while retaining or enhancing their biological activity.



Figure 1. Widely used aliphatic rings as bioisosteres to replace the phenyl rings

Among the aliphatic rings as bioisosteres of the phenyl rings, bicylco[1.1.1]pentane (BCP) attracts more attention of the medicinal chemists because of the comparable dihedral angle, the similar distance and the linear disposition of the substituents (**Figure 2**) with significantly improved properties of the compounds (**Figure 3**). 1,3-disubstituted BCP can mimic a parasubstituted phenyl ring.^[1]

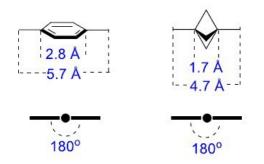
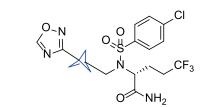


Figure 2. Geometrical parameters of phenyl ring and BCP ring

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V = N N F O = N O = S O = S O = CI CI CF_3 $O = NH_2$

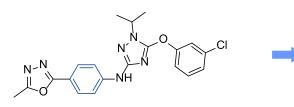
 $pIC_{50} = 9.65$ Solubility pH6.5 (kinetic) = 0.60 (uM) RRCK P_{app} (10⁻⁶ cm/s) = 5.52 RRCK CL (Human Hep.) = 15.0 (uL/min/10⁶ cells)



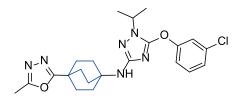
 $pIC_{50}=9.75$ Solubility pH6.5 (kinetic) = 216 (uM) $P_{app} (10^{.6} \text{ cm/s}) = 19.3$ CL (Human Hep.) = <3.8 (uL/min/10⁶ cells)

Figure 3. Effect on Potency and Developability Parameters of Bioisosteric Replacement of a phenyl ring of γ-Secretase Inhibitor with a BCP Moiety

The same effect was also observed on bicyclo[2.2.2]octane (BCO) ring system (**Figure 4**) and bridged piperidine ring system (**Figure 5**).^[1]

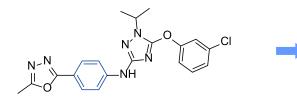


Cellular IC₅₀ = 510 (nM) Solubility (LYSA) < 0.1 (μ g/mL)

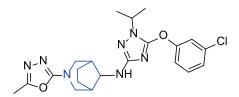


Cellular IC₅₀ = 118 (nM) Solubility (LYSA) = 15 (μ g/mL)

Figure 4. Effect on Potency and Developability Parameters of Bioisosteric Replacement of a phenyl ring of γ-Secretase Modulator with a BCO Moiety



Cellular IC₅₀ = 510 (nM) Solubility (LYSA) < 0.1 (μ g/mL)



Cellular IC₅₀ = 42 (nM) Solubility (LYSA) = 104 (μ g/mL)

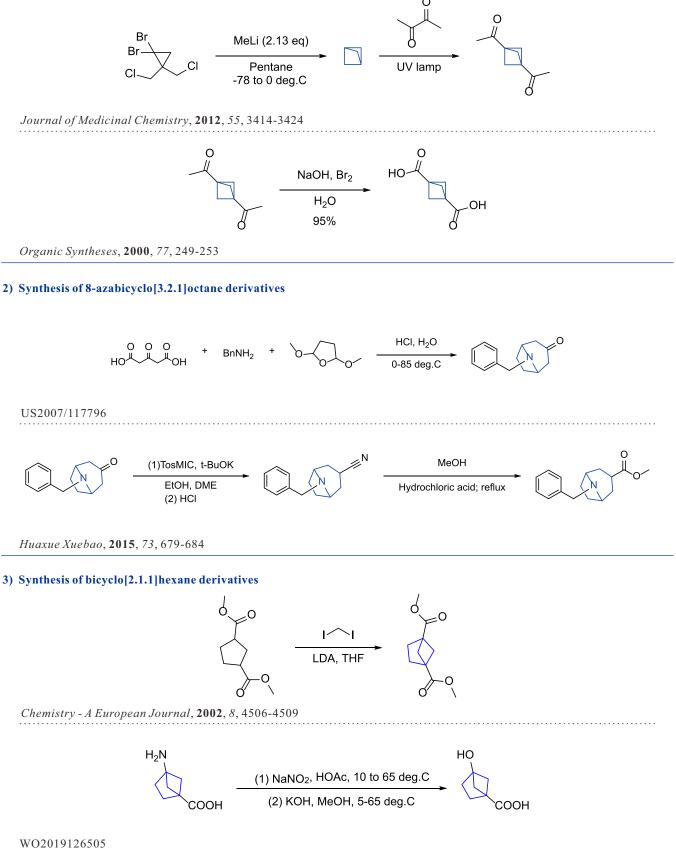
Figure 5. Effect on Potency and Developability Parameters of Bioisosteric Replacement of a phenyl ring of γ-Secretase Modulator with a Bridged Piperidine Ring System

Aliphatic rings have emerged as valuable bioisosteres of phenyl rings in medicinal chemistry because of their unique characteristics to modulate drug properties while retaining or enhancing improved stability, reduced toxicity, and enhanced pharmacokinetic profiles. Continued exploration and optimization of aliphatic ring substituents hold promise for the development of novel and effective therapeutics across various therapeutic areas.

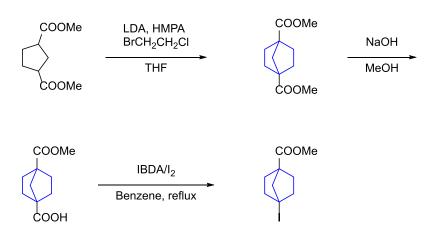
Whitepaper

* Synthesis of Aliphatic Rings



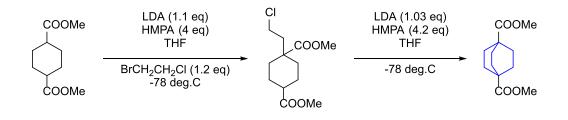


4) Synthesis of bicyclo[2.2.1]heptane derivatives



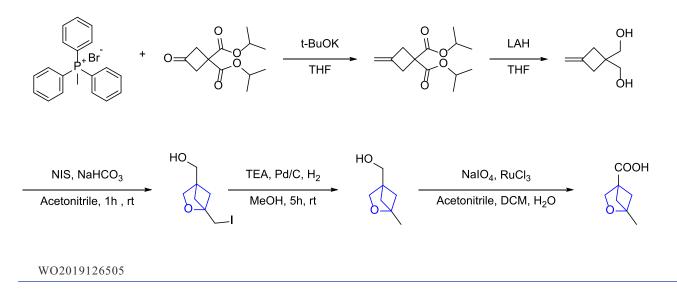
Synthetic Communications, 2007, 37, 1267-1272

5) Synthesis of bicyclo[2.2.2]octane derivatives



Journal of Medicinal Chemistry, 2011, 54, 3480-3491

6) Synthesis of 2-oxabicyclo[2.1.1]hexane derivatives



Building Blocks as Bioisosteres of Phenyl Rings

PharmaBlock has conducted a systematic study of clinical and preclinical drug molecules, and our chemists continue to pay attention to the latest research, design and synthesize a large number of bioisosteres of phenyl rings, which can be used to explore structure-activity relationship (SAR) and structure-property relationship (SPR). We offer more than 3000 unique bioisosteres of phenyl rings, ranging from grams to kilograms, most of which are in stock (**Figure 6**)

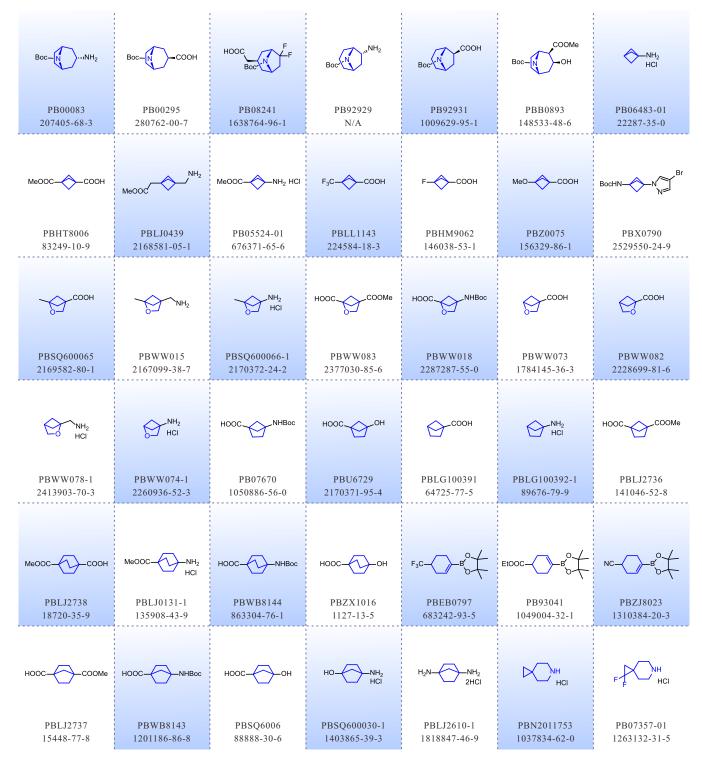


Figure 6. Representative bioisosteres of phenyl rings at PharmaBlock

References

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